

# Development of Alkaline Water Electrolyzer Technology at BARC for Green Hydrogen Production

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24<sup>th</sup> June, 2022



Seminar on Production & Use of Green Hydrogen and Green Ammonia in Process Industry

# GREEN HYDROGEN

**Solar**



**Wind**



**Hydel**



**WATER ELECTROLYSIS**

**GREEN HYDROGEN**

**BLACK  
HYDROGEN**

**GREY  
HYDROGEN**

**BLUE  
HYDROGEN**

**INDIA**

**H<sub>2</sub>: 6 million  
MT/yr**

**CO<sub>2</sub>: 60 million  
MT/yr**

**Natural Gas  
Import  
Dependency**

**% Import  
53 %**

**Import Value  
9.5 billion USD**

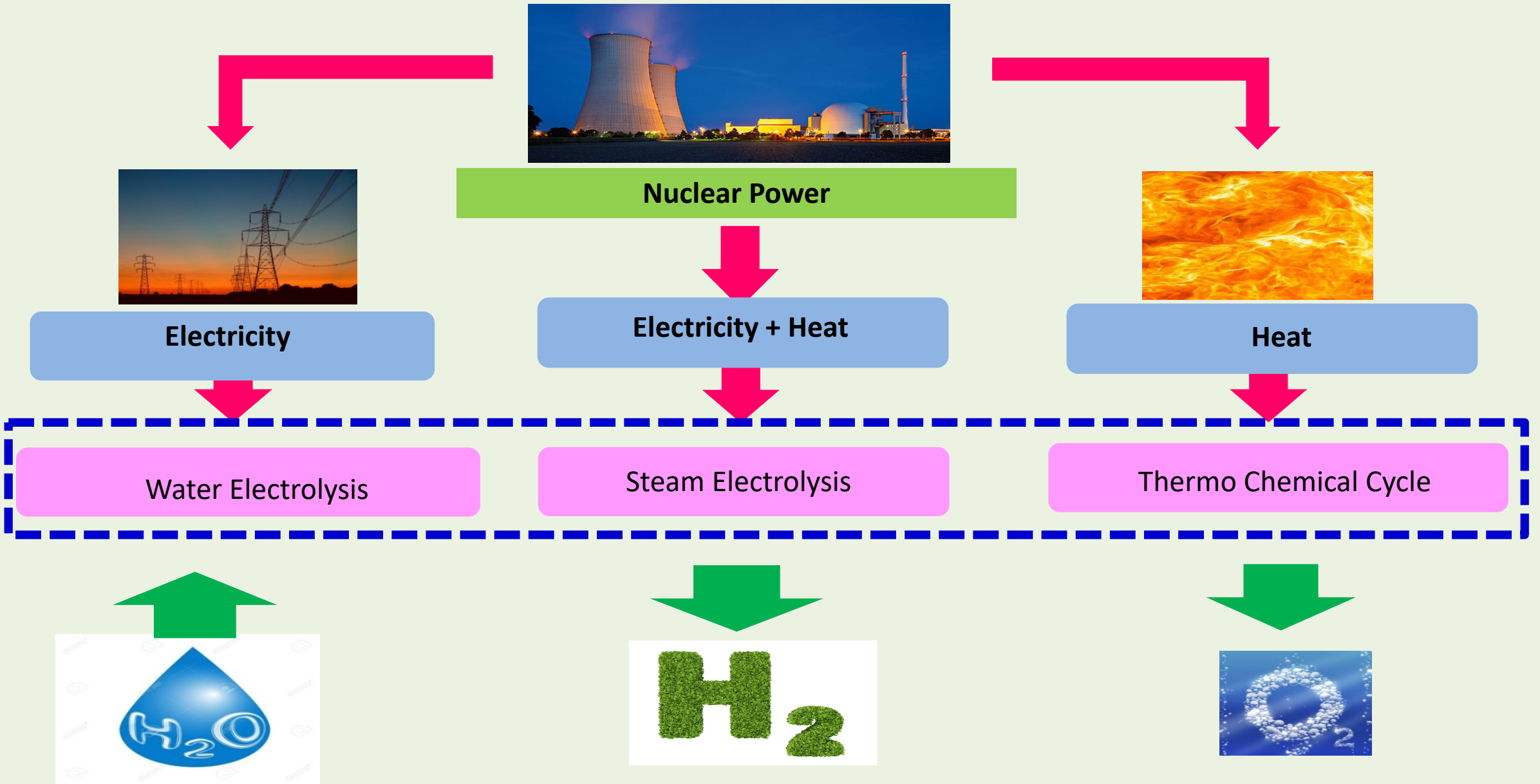
**Decarbonisation of  
sectors**

**Reduce Feedstock Import  
Bills**

**Address Climate  
Change Issue**

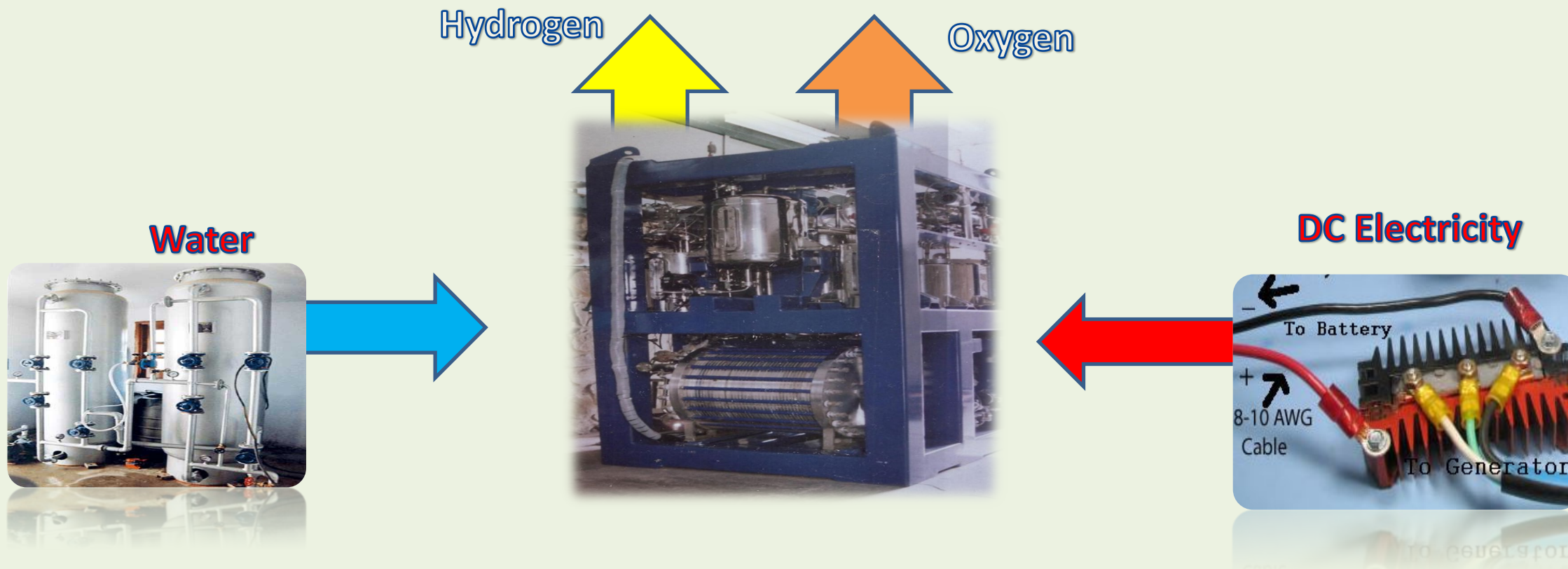
**Promote Energy Security & Self-  
Reliance**

# BARC Activities for water splitting



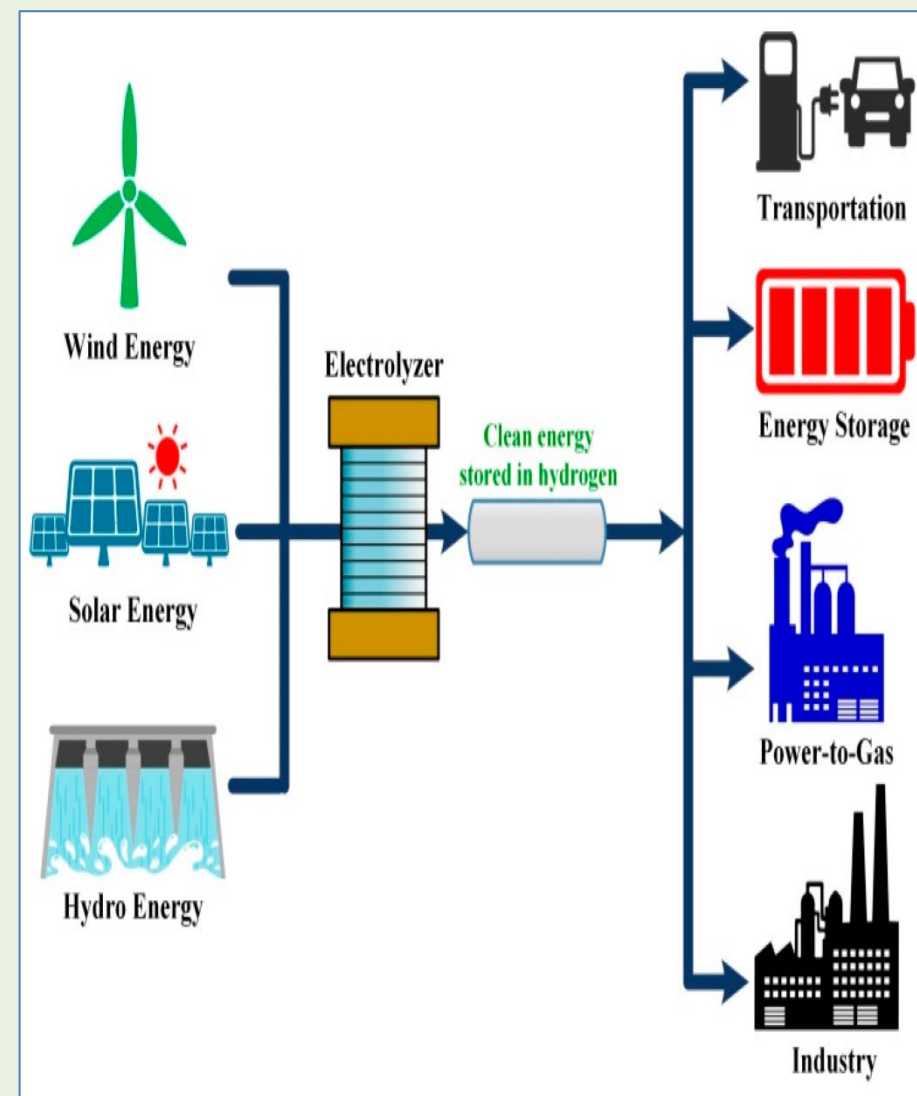


# LOW TEMPERATURE WATER ELECTROLYSIS



# Ideal Scenario for Low Temperature Water Electrolysis

- ✓ Produces high purity  $H_2$  & valuable  $O_2$  by product in a single step
- ✓ Operates at ambient conditions
- ✓ No  $CO_2$  emission
- ✓ High turn down ratio (20 to 100%)
- ✓ On demand, On site  $H_2$  production
- ✓ Centralized and decentralized source of hydrogen
- ✓ Modular Scalability
- ✓ Commercially matured; but imported.



# APPROACH

- Current density and cell voltage mainly depends up on the following

- Electrode material:-

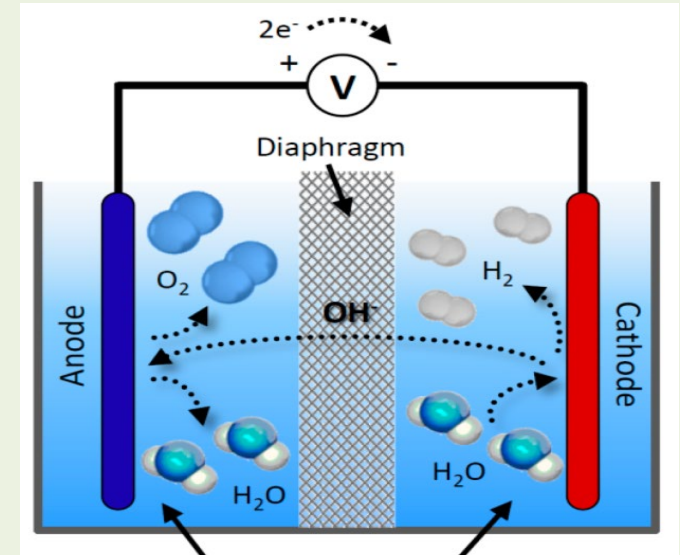
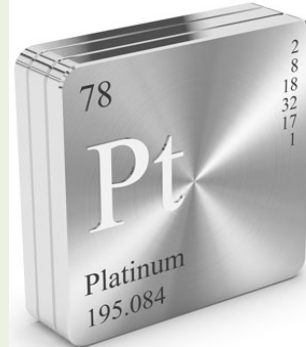
Steel



Nickel



Platinum



- Type of electrode configuration:



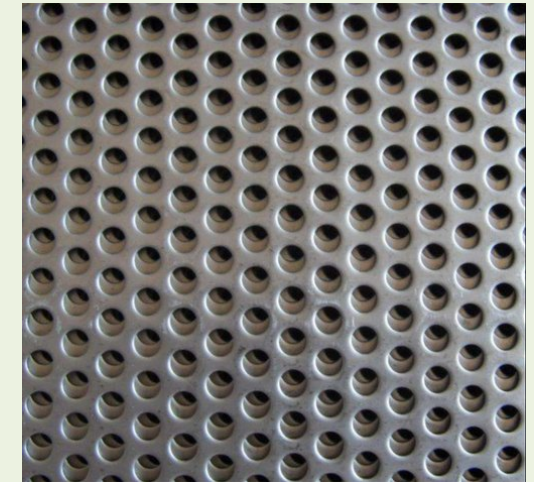
Plate Type



Extended surface



Porous electrodes

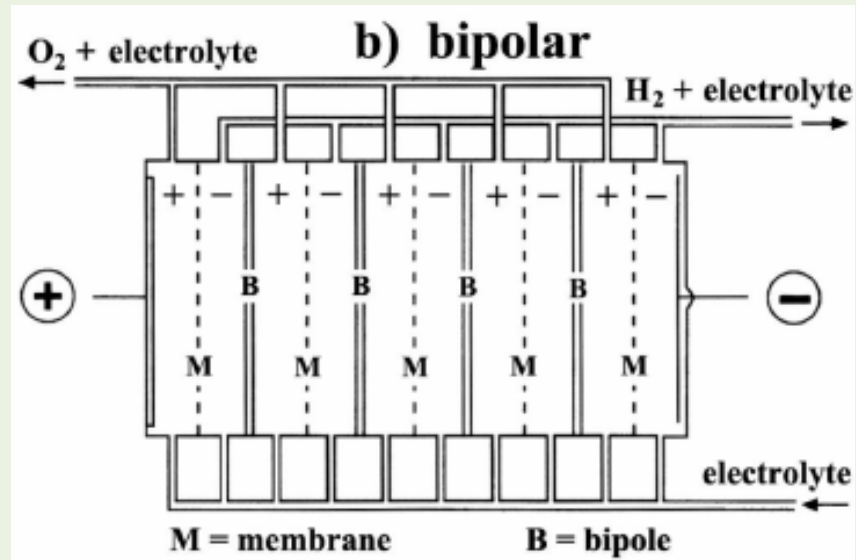


Perforated electrodes

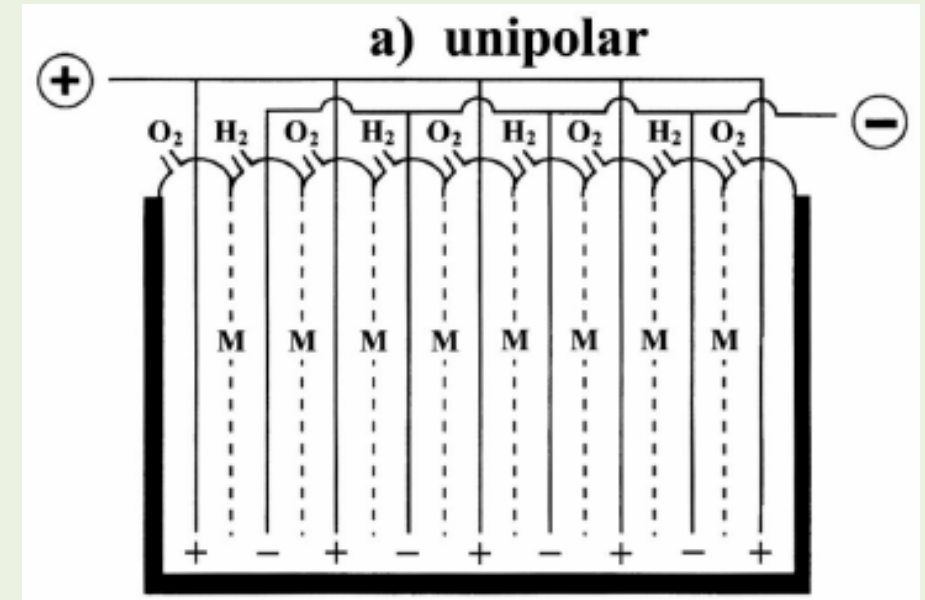


# APPROACH

## •Electrolyzer design : – Bipolar / Unipolar



Ref : <https://doi.org/10.1590/S0100-40422013000800017>



## •Electrolyzer design : Filter press type



Ref: <https://www.mvsengg.com/products/hydrogen/pure-hydrogen/>

## Tank Type



# APPROACH

## •Diaphragm materials: - Inorganic/ Organic/ Anion exchange



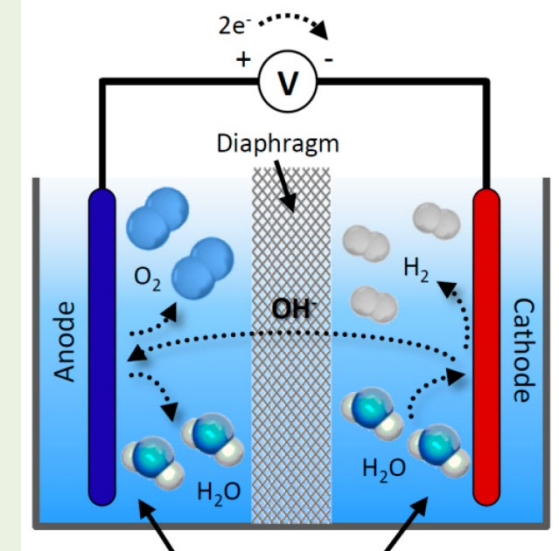
Asbestots



Polysulfones



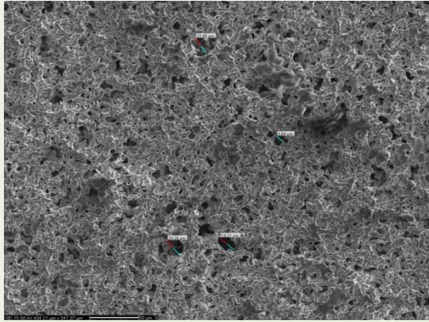
Anion exchange membrane



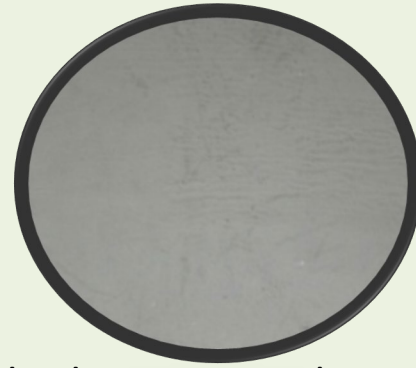
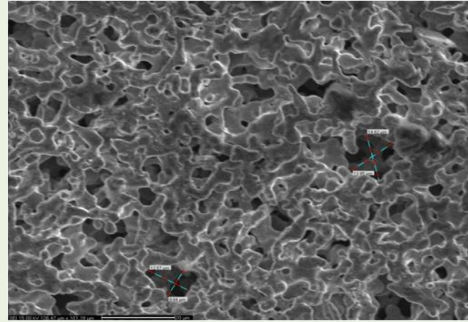
- ✓Nickel was a cost effective option with favourable kinetic properties and appreciable corrosion resistance towards alkaline medium
- ✓According to literature use of Porous Ni electrode increased current density to up 10000 ASM, thus making cell module more compact for portable applications
- ✓Bipolar design was selected to reduce ohmic losses and hence increase efficiency
- ✓Asbestos was used due to its high ionic conductivity leading to lower ohmic drop



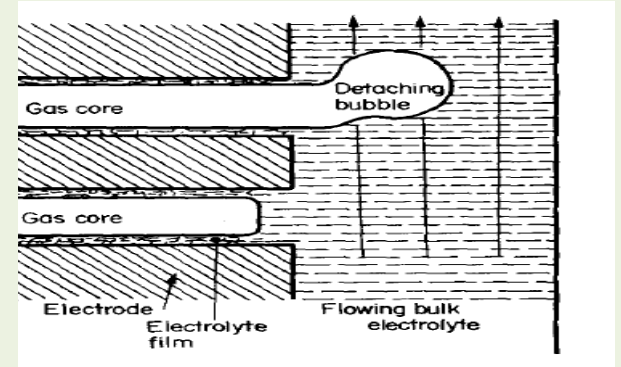
# POROUS NICKEL ELECTRODE DEVELOPMENT AT BARC



SEM Photograph of electrodes



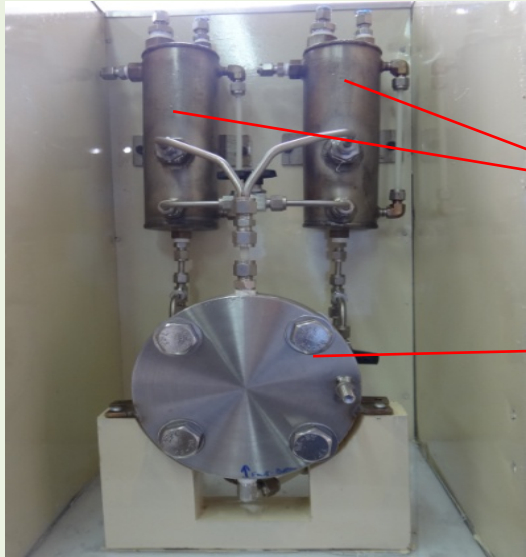
Finished Porous Ni electrode



Gas flow in porous electrode

- Heavy Water Division, Chemical Engineering Group, BARC is working on the development of porous Nickel electrodes
- Porous electrodes provide a large pore surface area compared to the cross sectional geometric area of the electrode.
- High internal pore surface area enhances overall reaction kinetics resulting in high current density
- Being highly porous ( $>70\%$  porosity), these electrodes can be sandwiched with the diaphragm (with zero gap) leading to a compact cell design.

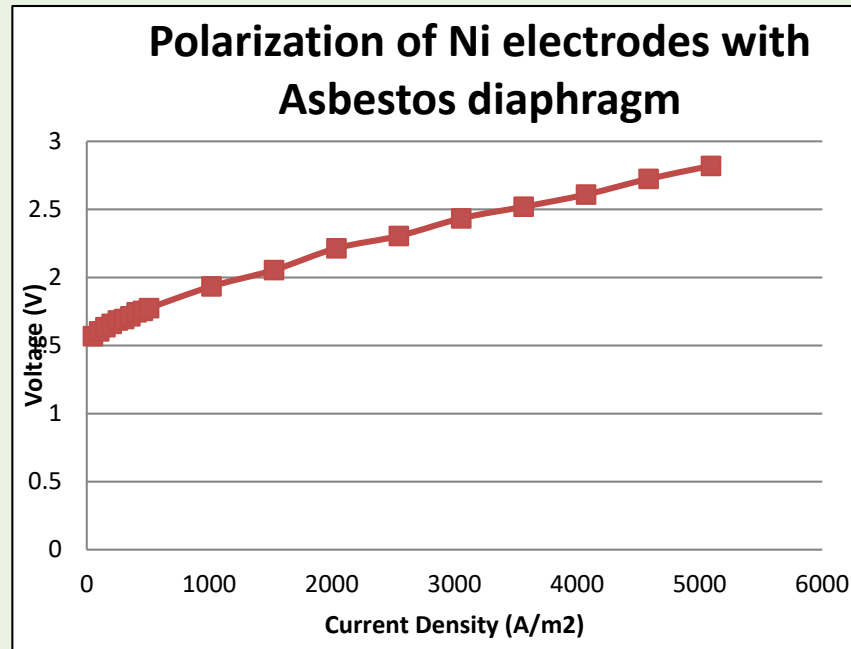
# LABS SCALE CELL MODULE DESIGN AND TESTING



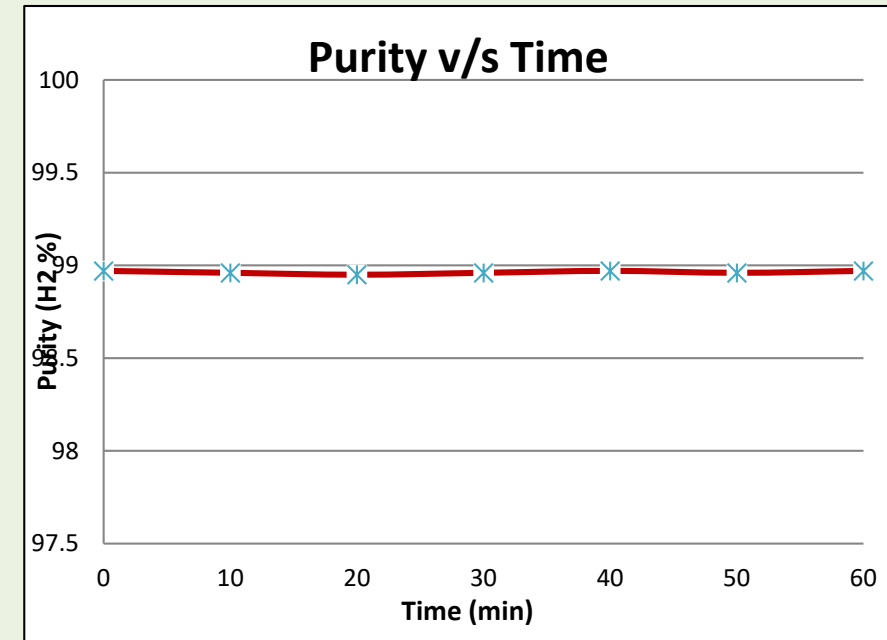
Gas separators

Cell module

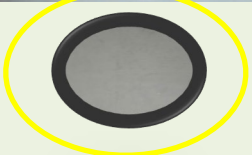
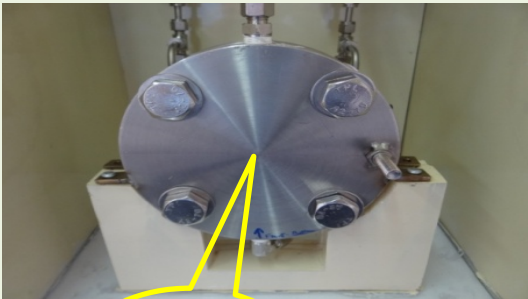
- The feasibility is established in a single cell lab scale electrolyzer with porous Nickel electrodes and asbestos diaphragm
- Hydrogen production rate : 4 Nlph



Polarization study

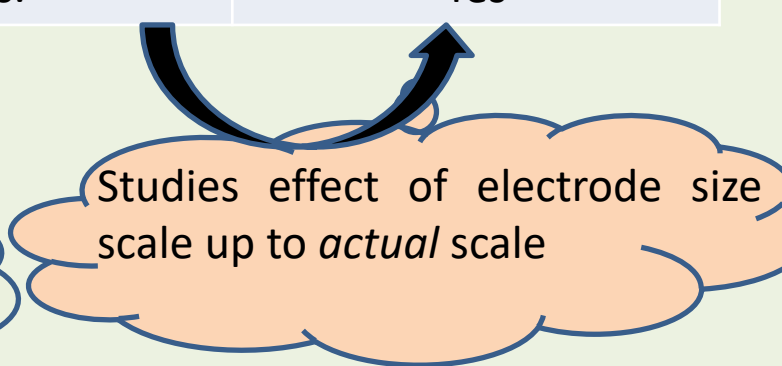
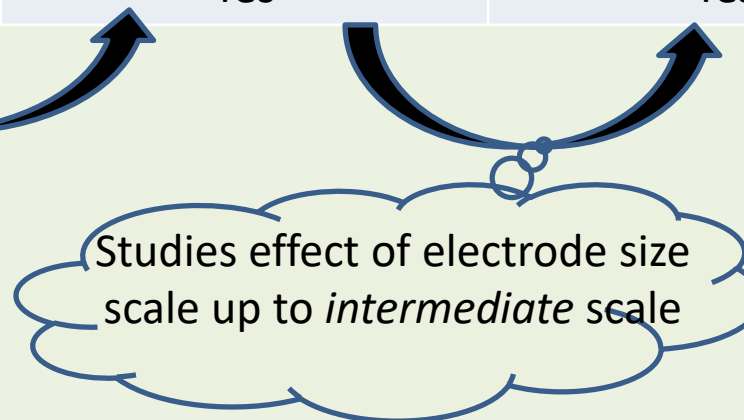
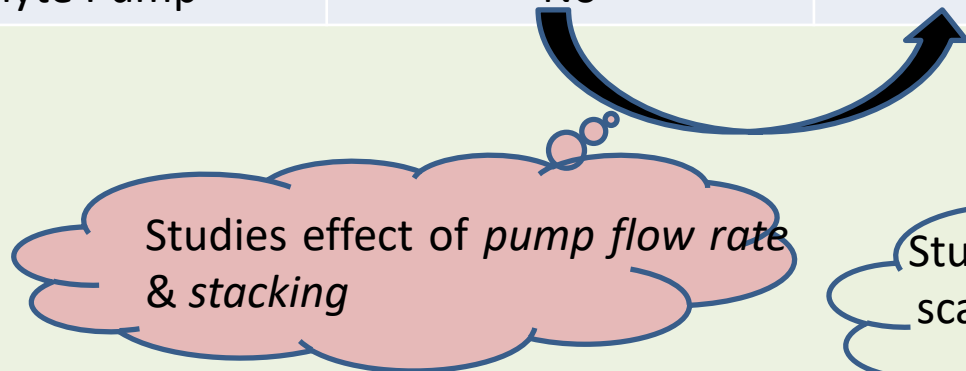


Effect of gas purity v/s Time



## FURTHER DEVELOPMENT AND SCALE UP

Feature	Lab scale	Lab scale( stack)	Bench Scale	Pilot Scale
Electrode Diameter scale up ratio	1	1	3	9
H <sub>2</sub> production rate	4 Nlph	12 Nlph	80 Nlph	1200 Nlph
Operating Pressure	Atmosphere	Atmosphere	Atmosphere	Up to 1.5 bar(a) maximum
Stacking	No	Yes	Yes	Yes
Electrolyte distributors	No	No	Yes	Yes
Electrolyte Pump	No	Yes	Yes.	Yes





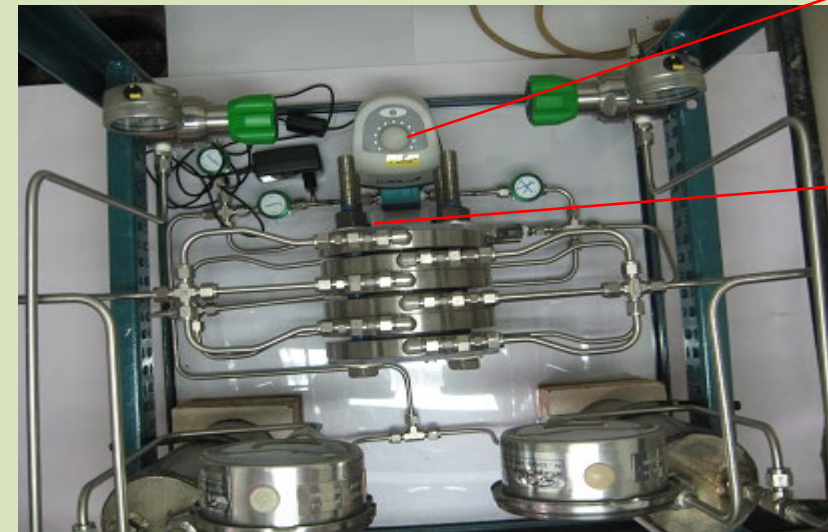
# SCALE UP TO LAB SCALE STACK ( 3 CELL MODULE)

Feature	Lab scale( stack)
Electrode diameter Scale up ratio	1
H <sub>2</sub> production Capacity	12 NLph
Operating Pressure	Atmosphere
Stacking	Yes (3 cell stack)
Electrolyte distributor	No
Electrolyte circulation	Yes



Front view

Gas separators  
Cell module



Top view

Pump  
Cell module

## SCALE UP TO BENCH SCALE STACK ( 2 CELL MODULE)



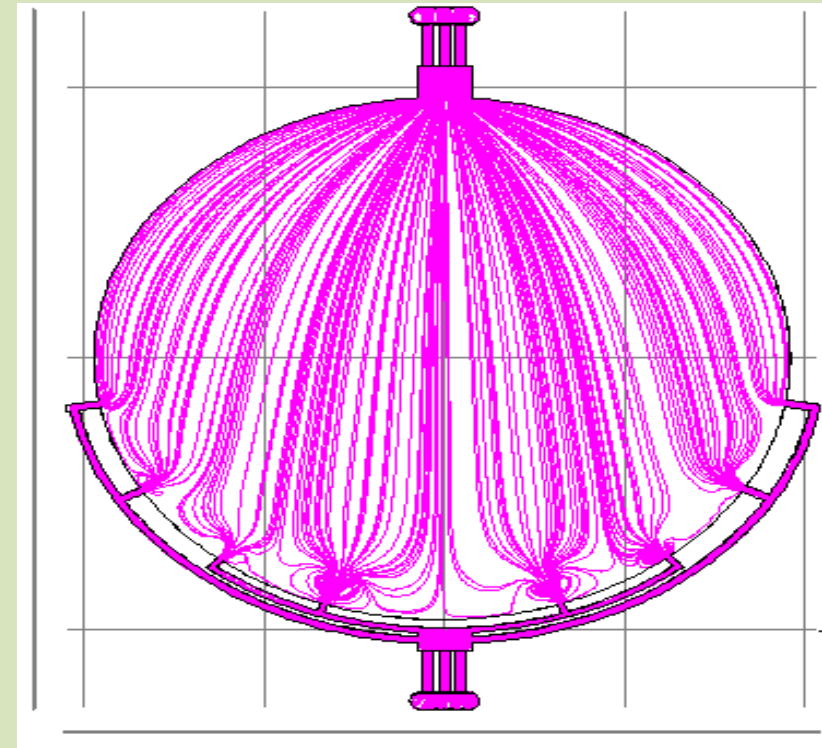
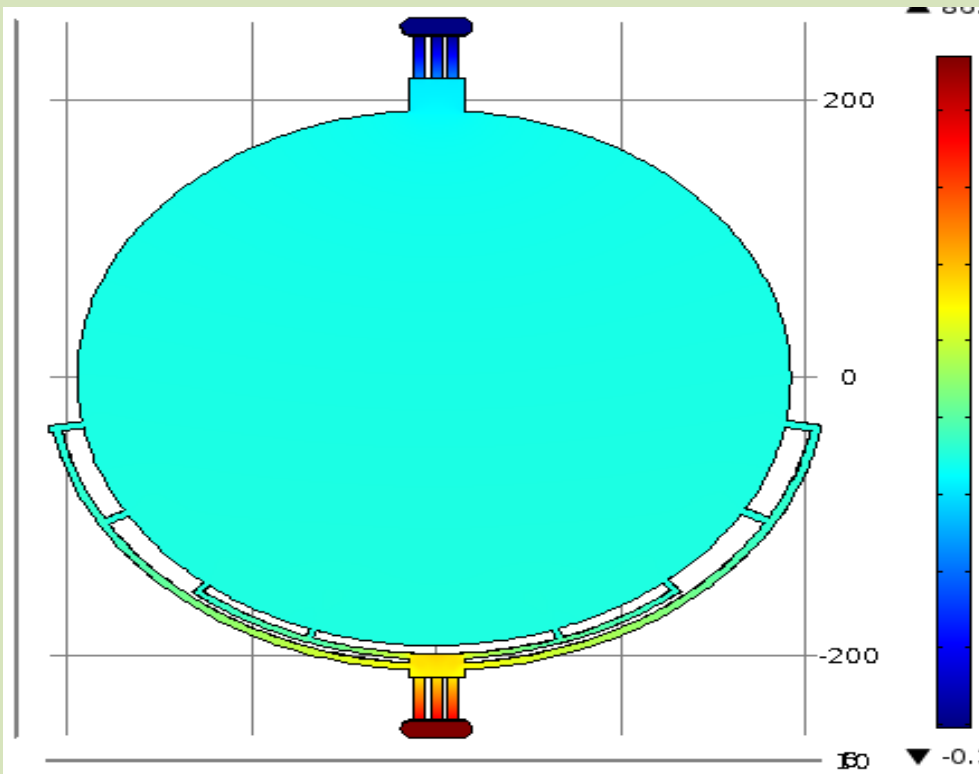
Gas separators

2 Cell module

Feature	Bench Scale
Electrode diameter Scale up ratio	3
H <sub>2</sub> production Capacity	80NLph
Operating Pressure	Atmosphere
Stacking	Yes
Electrolyte distributor	Yes
Electrolyte circulation Pump	Yes

# Preliminary CFD studies for electrolyte distribution through cell

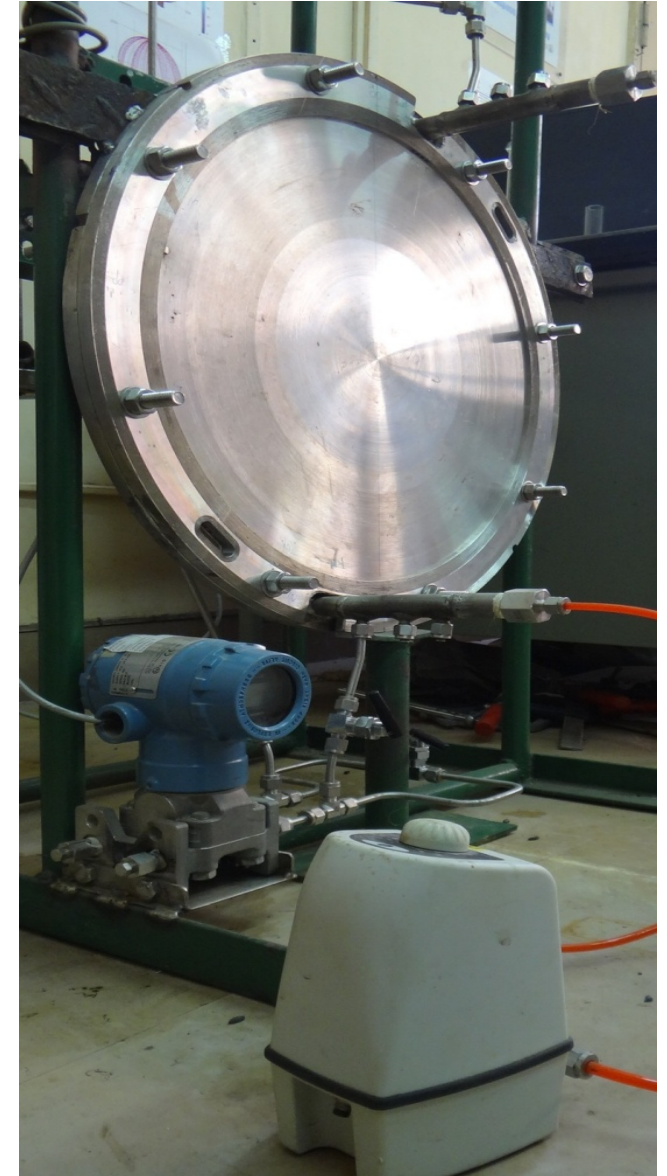
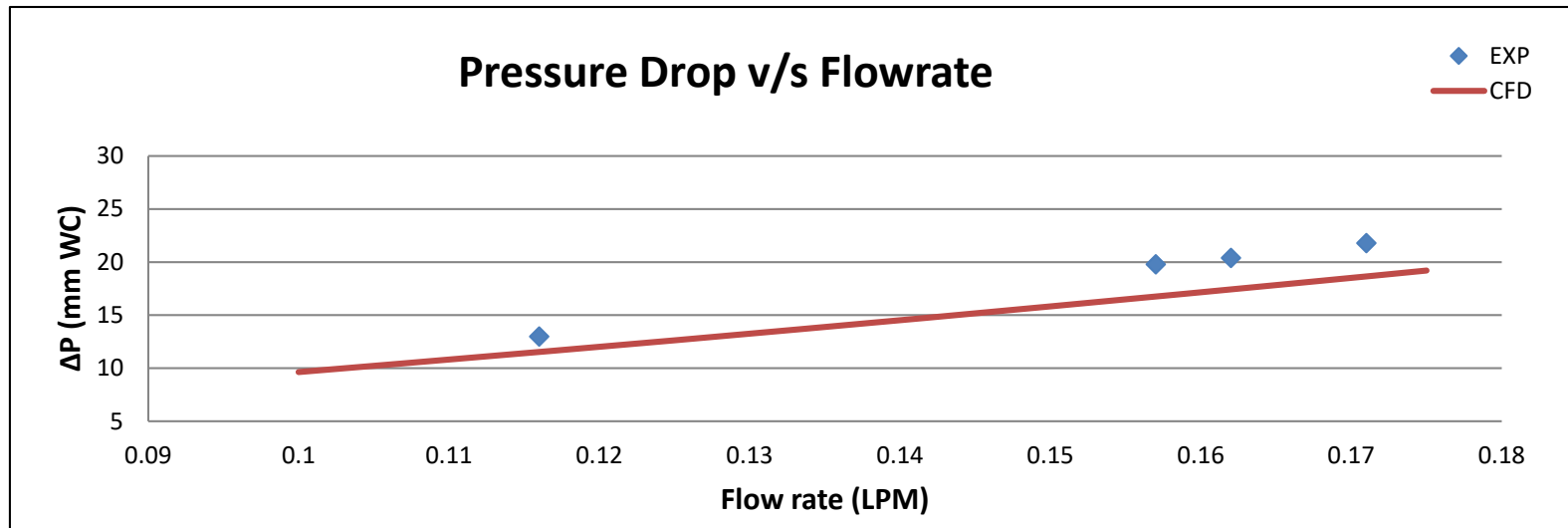
- Aim: To evaluate uniform distribution of electrolyte in the anodic/ cathodic flow channels
- Approach: Laminar Single phase electrolyte flow
- Future Work Plan: Modeling Two phase flow and validation by experiments
- To ensure proper uniform electrolyte distribution at min.  $\Delta P$  by trying different design modifications



*Pressure profile and Velocity streamline*

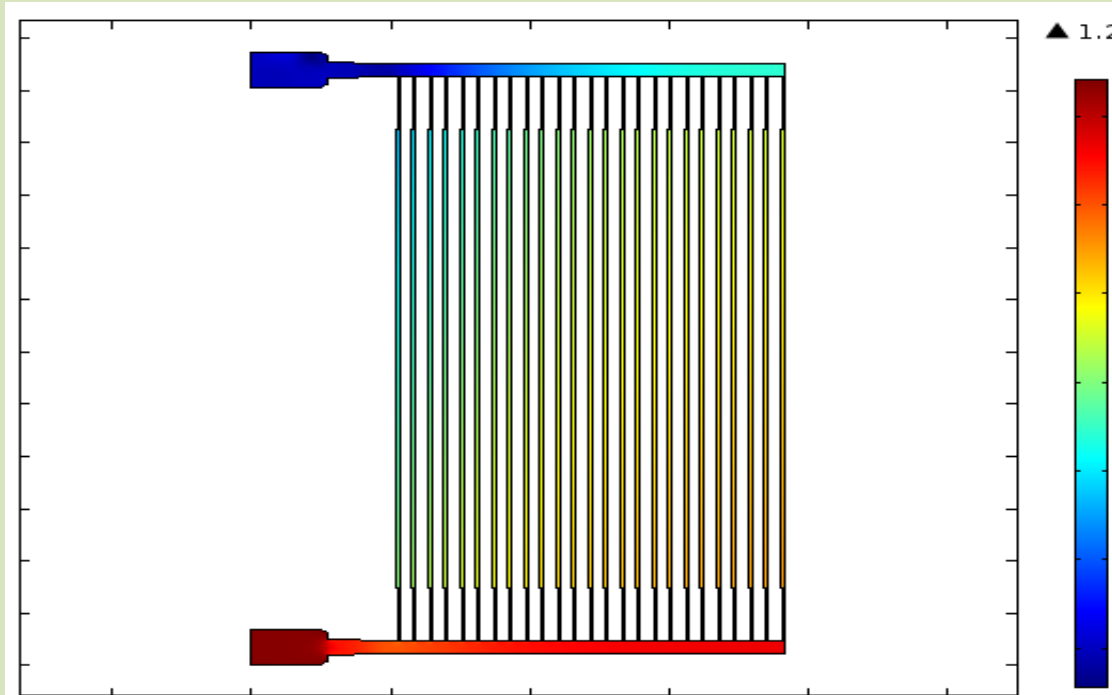


# Half Cell Hydro-dynamics Set Up

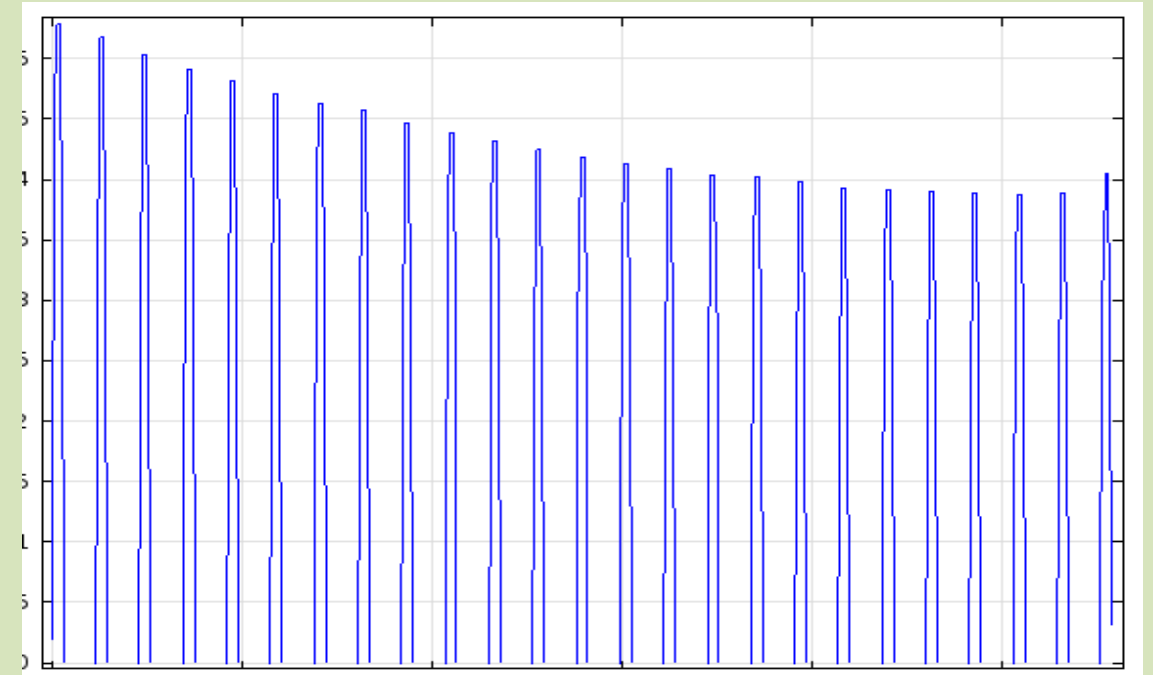


# Preliminary CFD studies in 25 cell electrolyser stack

- Aim: To determine  $\Delta P$  and velocity profile across the cell stack
- Approach: Laminar Single phase electrolyte flow
- Future Work Plan: Modeling Two phase flow and validation by experiments
- To ensure uniform thermal hydraulics in each channel by gradient design approach
- To arrive at min. total  $\Delta P$  for the cell stack by trying different designs



**Pressure Profile**



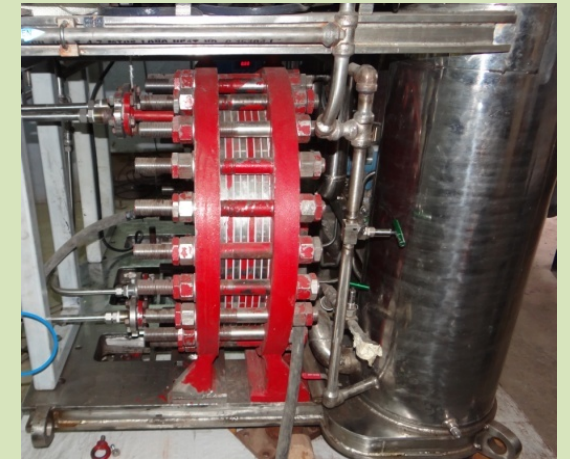
**Velocity Profile**

# PILOT SCALE CELL MODULE (6 CELL MODULE)

Feature	Pilot Scale
Electrode diameter Scale up ratio	9
H <sub>2</sub> production Capacity	1200 Nlph
Operating Pressure	Up to 1.5 bar(a) maximum
Stacking	Yes
Electrolyte distributor	Yes
Electrolyte circulation Pump	15 lpm max.



Process skid



Six Cell module

Gas separators



# DIFFERENT STAGES OF DEVELOPMENT – “LAB TO LAND”

Single Cell



H<sub>2</sub> :4 lph

Three Cell



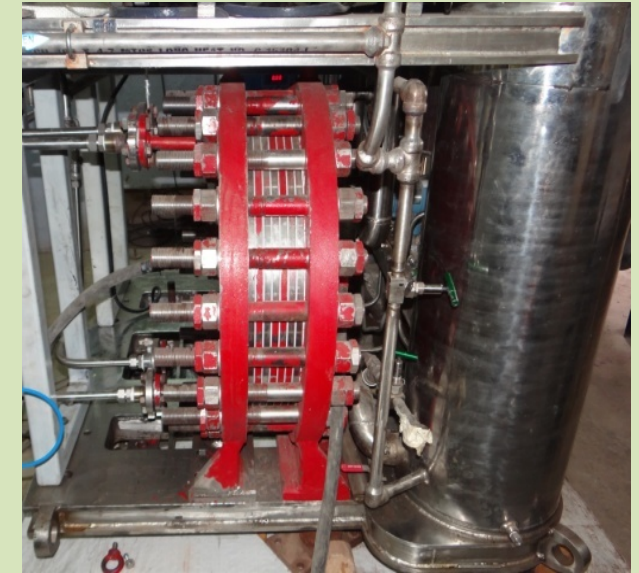
H<sub>2</sub> : 12lph

Two Cell



H<sub>2</sub> : 80 lph

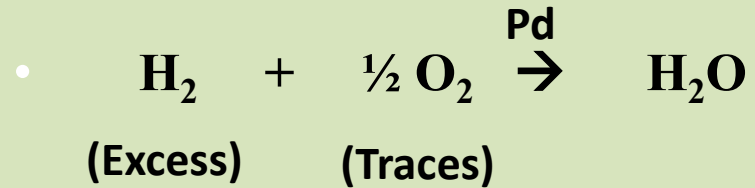
Six Cell



H<sub>2</sub> :1200 Nlph

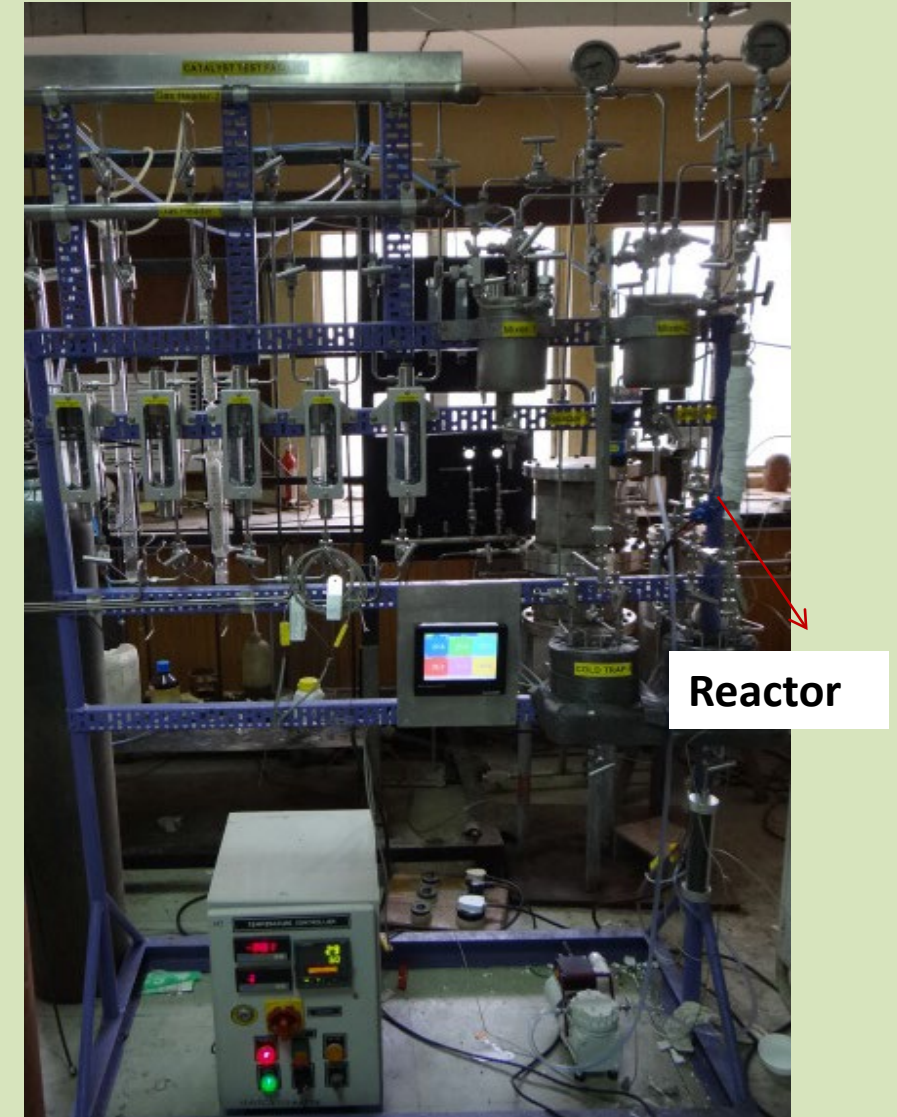
# H<sub>2</sub> GAS PURIFICATION USING PALLADIUM BASED CATALYST

•The final gas purification of the hydrogen gas is done using palladium based catalyst.



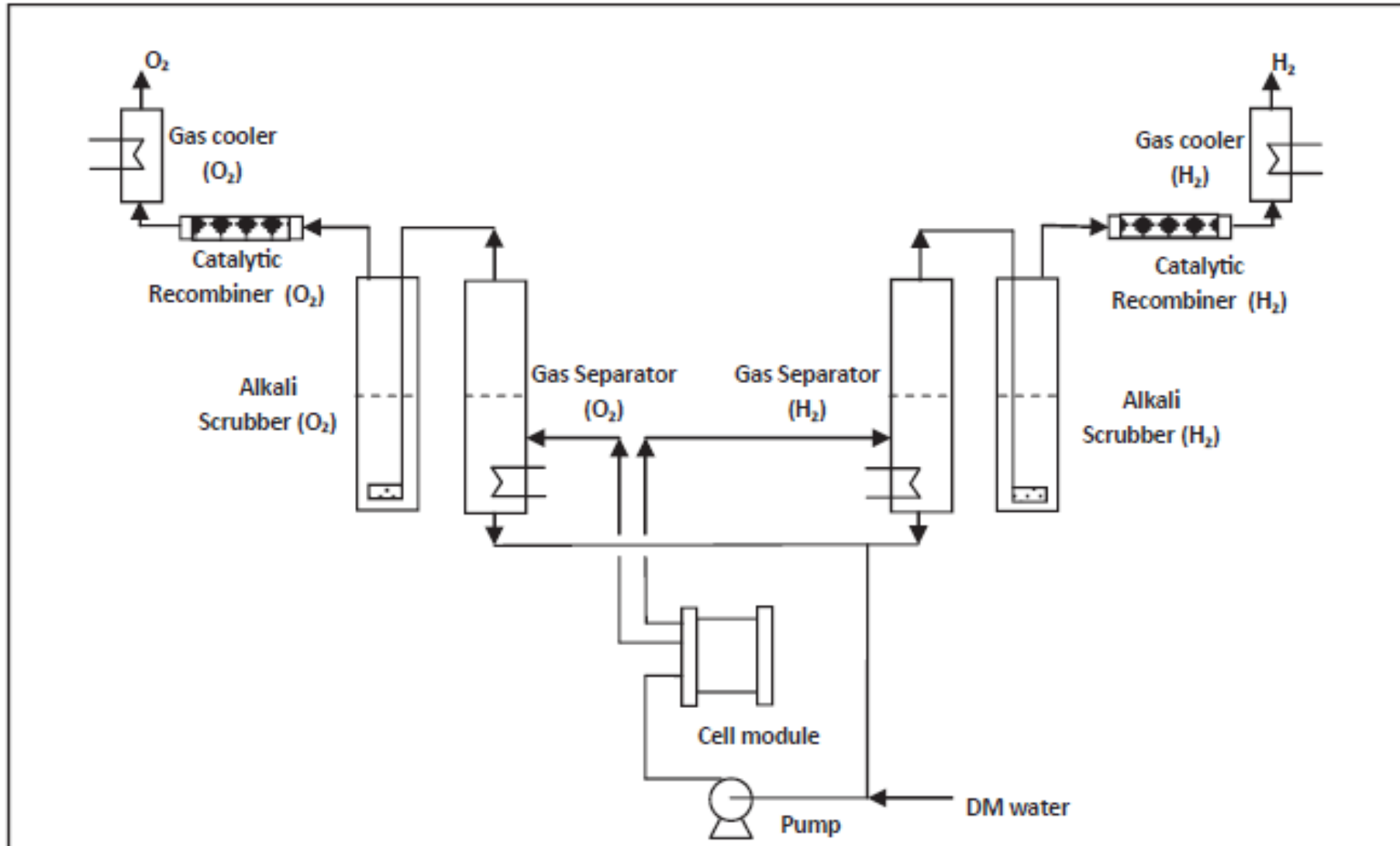
Different Catalysts developed with Chemistry Division, BARC

•In the electrolyzer plant, the final oxygen gas purity of >99.99% is achieved using this catalyst at plant exit.



Catalyst Test Facility

## SCHEMATIC OF SCALE UP TO 10 NM<sup>3</sup>/HR HYDROGEN PLANT





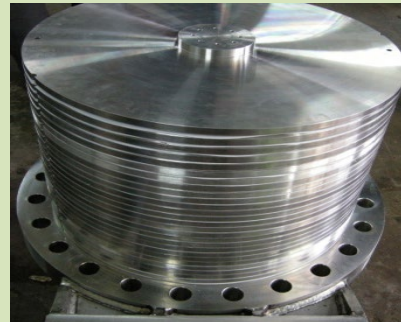
# DIFFERENT STEPS IN POROUS NICKEL ELECTRODE DEVELOPMENT

## ■The major steps involved are

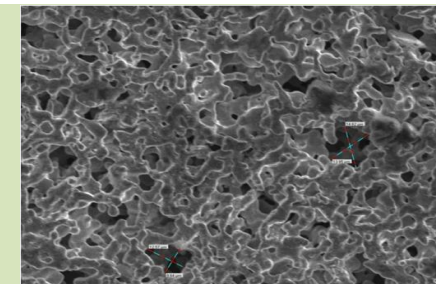
- 1) Fabrication of the green plaques by spray coating of Carbonyl nickel powder slurry on Ni wire mesh .
- 2) Roller compaction to desired thickness.
- 3) Sintering in hydrogen furnace at high temperature
- 4) EB welding of porous nickel sheet to the electrodes holder Rings



Cell modules after assembly

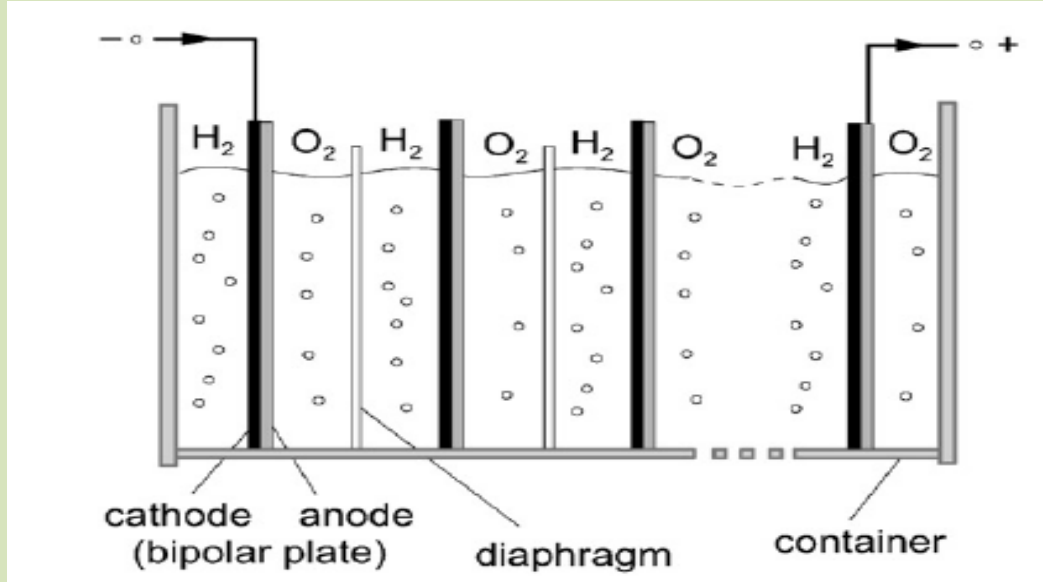


Cell module assembly



Porous Ni Electrode

# CELL MODULE STACK DESIGN



## ***Features:***

- Bipolar Compact filter-press type cell module design
- Thin sintered porous nickel electrodes (1mm)
- High current density (5000 ASM) leading to lower cell voltages
- Cell assembly with minimum gap (1 mm)
- High Purity product gases

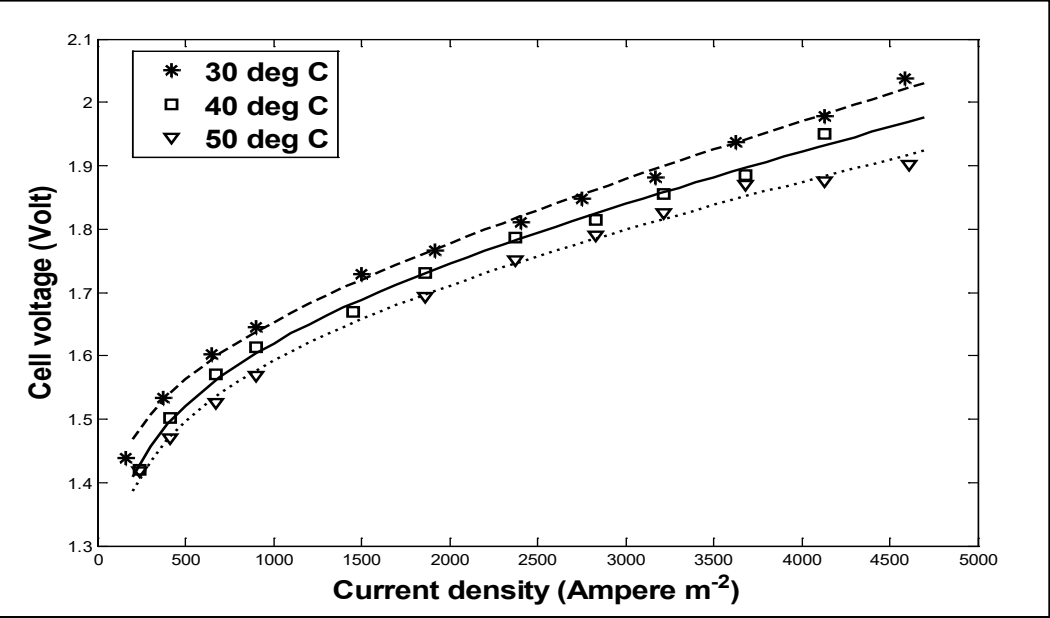
# First Production plant at BARC



- Indigenously developed zero gap, compact alkaline water electrolyser technology
- On demand, On site, High purity Hydrogen & Oxygen produced
- Safe, reliable and robust plant design and automated control philosophy
- Successful operational experience of over 5000 hrs with high availability factor
- Four Nos. of units each capable of producing hydrogen up to 10000 Nlph have been developed and deployed



# ELECTROCHEMICAL PERFORMANCE EVALUATION OF ELECTROLYZER PLANT (10000 NLPH H<sub>2</sub>)



Comparison of Ohmic parameter with literature ( Kibria et al 2011)

Temperature  (°C)	Ohmic parameter (b)  ( m <sup>2</sup> )	
	Reported	BARC
30	7.3 x 10 <sup>-5</sup>	6.81x10 <sup>-5</sup>
40	7.05 x 10 <sup>-5</sup>	5.32x 10 <sup>-5</sup>
50	6.8 x 10 <sup>-5</sup>	4.61x10 <sup>-5</sup>

•The operational cell voltage versus current density is generated at different operating temperatures ( 30, 40 & 50 deg C)

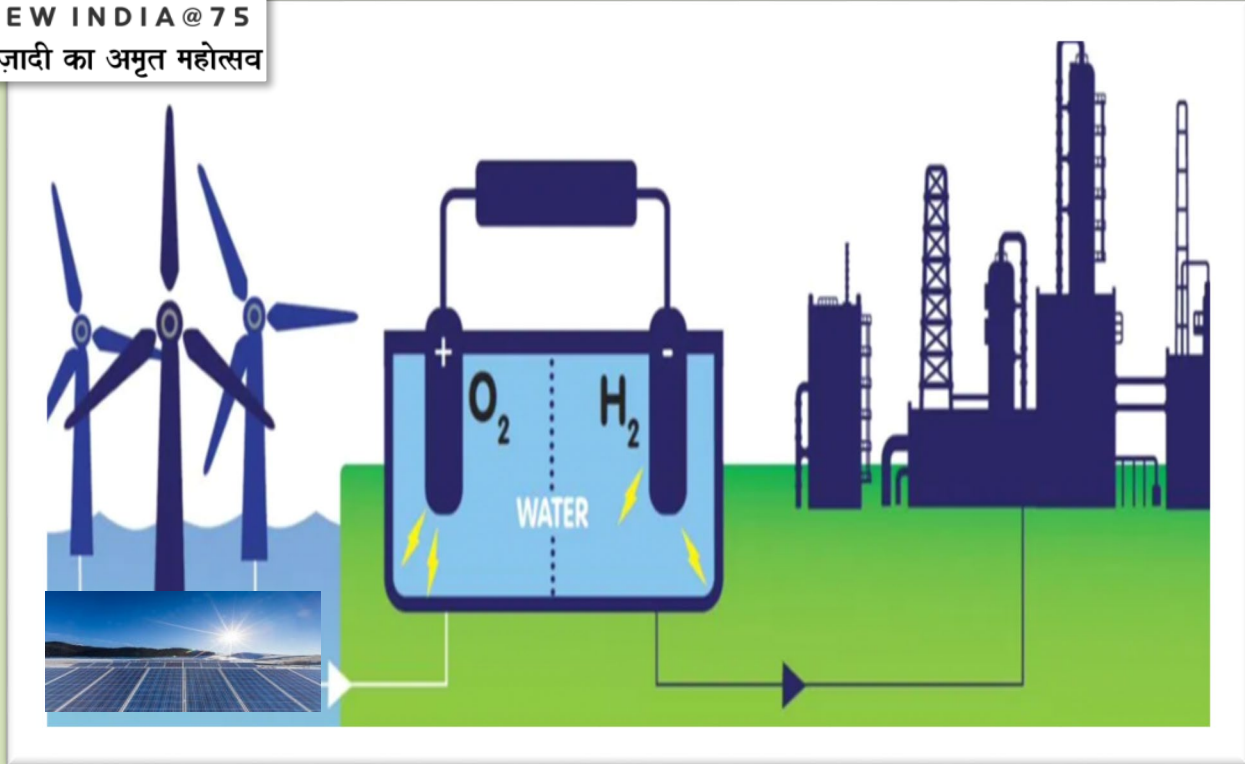
•The different model parameters accounts for overall cell resistance such as Ohmic & Activation over potentials were estimated.

Comparison of exchange current density (Am<sup>-2</sup>) with literature ( Ullsberg et al 2012)

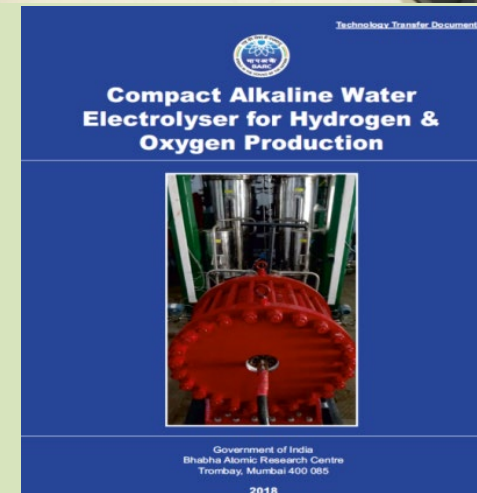
Temperature  (°C)	Exchange current density  (A m <sup>-2</sup> )	
	* Reported for Ni at 30% KOH	BARC
30	6.9540	12.0495
50	18.3251	43.8657



# TECHNOLOGY TRANSFER TO INDUSTRY



Technology transferred to HPCL in 2021





# TECHNOLOGY TRANSFER TO BPCL

## BPCL ties up with BARC for electrolyser technology to produce green hydrogen

### Bharat Petroleum collaborates with BARC for green Hydrogen production

BY PSU CONNECT · DATE: 13-12-21

Bharat Petroleum has plans to expand its portfolio of renewable energy with solar, wind and biofuels thereby reaffirming its commitment towards sustainability and reduction of carbon footprint.



AGREEMENT SIGNING AT BARC, MUMBAI



## Indian refiner BPCL tie-up with nuclear institute in net zero push



India's third largest refiner on Monday it has tied up with the country's nuclear research institute for an electrolyser technology to produce green hydrogen as it seeks to achieve net zero emissions by 2040. For green hydrogen production, renewable energy is used for electrolysis.



## BPCL teams up with BARC

DH DECCAN HERALD Home Coronavirus Bengaluru Karnataka National Sports Business Opinion Features Videos

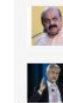
### BPCL ties up with Bhabha Atomic Research Centre in net zero push

BPCL's Chairman A.K. Singh said the refiner plans to scale up indigenous alkaline electrolyser technology for use in refineries

Reuters, New Delhi, DEC 13 2021, 18:45 IST | UPDATED: DEC 13 2021, 18:45 IST



DH PICKS



### GREEN GOALS

India, the world's third biggest greenhouse gas emitter, is aiming to reach net zero emissions by 2070 and wants to raise the share of renewables in its energy mix to 50 per cent by 2030 from 38 per cent at present.

Refineries use large quantities of hydrogen for desulfurisation to make petrol and other chemicals. At present, this hydrogen is produced using steam reforming of natural gas, but results in CO2 emission. Therefore, refineries are planning to set up large-scale electrolysers to produce green hydrogen from water and decarbonise hydrogen production.

"Through collaboration with BARC, we intend to scale up indigenous alkaline electrolyser technology and look forward to commercialising it for large use, especially in Refineries," BPCL's chairman said.

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Home / Industry / Energy / BPCL and Bhabha Atomic Research Centre collaborate for green hydrogen production

### BPCL and Bhabha Atomic Research Centre collaborate for green hydrogen production



### BPCL TIES UP WITH BARC FOR GREEN HYDROGEN



NIDHI VERMA  
NEW DELHI, DEC. 13

India's third largest refiner Bharat Petroleum Corp Ltd said on Monday it has tied up with the country's nuclear research institute for an electrolyser technology for green hydrogen as it seeks to achieve net zero emissions by 2040.

India, the world's third biggest greenhouse gas emitter, is aiming to reach net zero emissions by 2070 and wants to raise the share of renewables in its energy mix to 50 per cent by 2030 from 38 per cent currently.

BPCL's chairman A.K. Singh said the refiner plans to scale up indigenous alkaline electrolyser technology for use in refineries via collaboration with the Bhabha Atomic Research Centre.

An electrolyser splits water into hydrogen and oxygen using electricity. For green hydrogen, renewable energy is used for electrolysis. At present Indian companies import electrolyser plants.

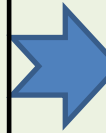
India's draft National Hydrogen Mission mandates that refiners and fertiliser producers should meet half of their hydrogen needs through green hydrogen by 2030. The central government has provided financial support for setting up electrolysers in the country.

Companies, including Reliance Industries, Adani Group, Indian Oil Corp, NTPC and GAIL have announced plans to play a critical role in India's green hydrogen road map.

Refineries use large quantities of hydrogen for de-sulfurisation to make petrol and other chemicals. —Reuters



# Spin off : “Medical Oxygen during pandemic”



- Capable of producing ~5000 Litre O<sub>2</sub> in 1 hr
- No logistic issues
- Capable of storing 6000 N litres O<sub>2</sub>
- O<sub>2</sub> stored under high pressure: 120 bar
- O<sub>2</sub> purity >99.9%



# BARC joins Covid Fight



**All India Radio News**   
@airnewsalerts

Bhabha Atomic Energy Research Centre sets up a special [#oxygen](#) plant to supply medical oxygen to Mumbai.

7:38 PM · 06/05/21 · [TweetDeck](#)

← Rahul Shewale - राहुल शेवाळे

BARC scientists were appealed on my behalf to solve the problem of oxygen deficiency in Mumbai. Giving a positive response to it, a water-producing oxygen plant has been implemented by BARC. In the first phase from this plant, about 10 oxygen cylinders will be supplied each of 50 liters. This can solve the problem of oxygen in government hospitals, Covid centers in south-central Mumbai. Along with the supply of oxygen, BARC has developed a special mask and remote body temperature machine for the citizens. I #thank the scientists and officers of #BARC who are rushing to help Mumbai during the Corona crisis! #FightAgainstCovid

#FightAgainstCoronavirusTogether  
Bhabha Atomic Research Centre Bhabha Atomic  
Research Centre Bhabha Atomic Research Centre  
India Bhabha Atomic Research Centre, Mumbai  
CMOMaharashtra ShivSena Shivsena MyBmc

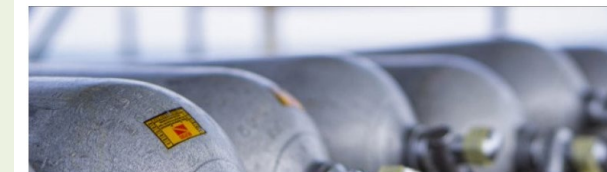
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## COVID-19 Pandemic: BARC To Supply Oxygen Cylinders To Mumbai

According to sources, the cylinders will be supplied to government hospitals and COVID-19 care centres in south-central Mumbai.

BY [MUMBAI LIVE TEAM](#) • 7 DAYS AGO • [HEALTH](#) • [MUMBAI](#)



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BARC Provides Oxygen Supply To  
Mumbai: BARC to supply oxygen to  
Mumbai; Information of MP Rahul  
Shewale  
BARC Provides Oxygen Supply  
To Mumbai: BARC to supply oxygen to  
Mumbai; Information of MP Rahul  
Shewale  
BARC Provides Oxygen Supply  
To Mumbai: BARC to supply oxygen to  
Mumbai; Information of MP Rahul  
Shewale



# Proposed Scale Up Strategy

2 x 50 cells



Increase Stack Length



Increase Stack Diameter



$H_2$ : 30 Nm<sup>3</sup>/hr

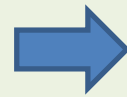
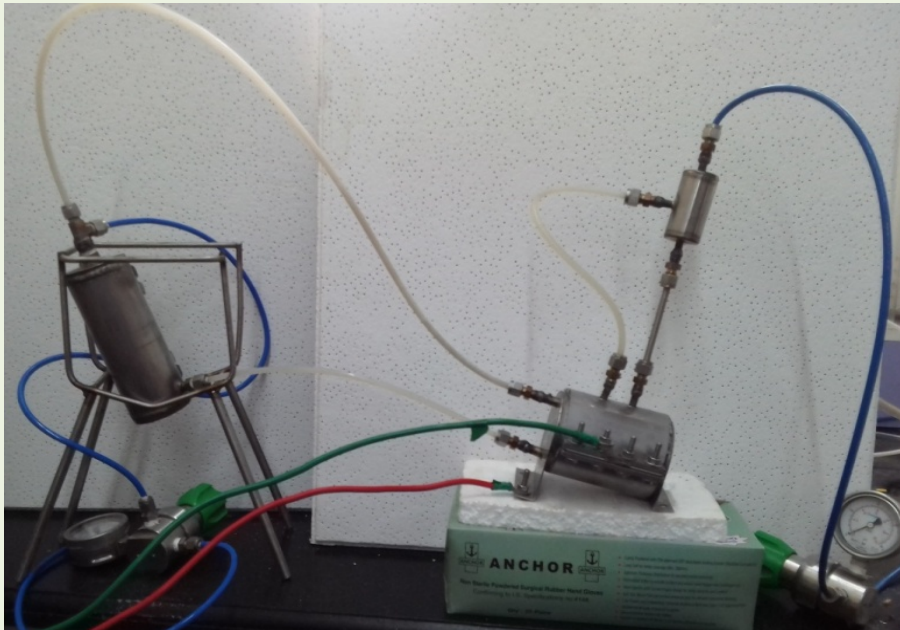
120 Nm<sup>3</sup>/hr

500 Nm<sup>3</sup>/hr



# Development of Lab Electrolyser

- Compact, lab scale unit capable of delivering 4 NLPH High Purity  $H_2$  &  $O_2$
- Both  $H_2$  &  $O_2$  are delivered at pressure up to 8 bar
- Ideal for small lab scale high purity Carbon free applications
- Easy control & plug on system
- Target to increase production up to 18 NLPH
- Product aesthetics



# Acknowledgment

- Dr. R R Sonde, IIT Delhi
- Desalination & Membrane Technology Division, BARC
- Fire Service Section, BARC
- Safety Review Committee & BSC (SF), BARC
- Chemistry Division, BARC
- Chemical Technology Division, BARC
- Alkali Materials & Metals Division, BARC



# *Team*







Let's together bring a difference..

# THANK YOU

